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(54) Title: ALKYL GLUCOSIDE, ITS USE FOR CLEAN	ING P	URPOSES, AND CLEANING COMPOSITION
(57) Abstract		
In an alkyl glucoside of the formula (I): RCH2O(G)xF	ł, R is	an alkyl group having a total of 8-12 carbon atoms and containing 2-4

groups of formula -CH(CH₃)- in its carbon chain; G is a monosaccharide residue; and x is 1-4. The use of the alkyl glucoside as a surfactant in the cleaning of hard surfaces is also disclosed. A composition containing an alkyl glucoside of formula (I), a complexing agent and a solubiliser is further described."

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ALKYL GLUCOSIDE, ITS USE FOR CLEANING PURPOSES, AND CLEAN-

This invention relates to an alkyleglucoside in which the alkyleglucoside in which the alkyleglucoside as methyl-branched mas well asother use of the alkyleglucoside as a surfactant in the releaning of hard surfaces. The invention also concerns accleaning composition in which the alkyleglucoside rist combined with a solubiliser and preferably also with a solubiliser and preferably also with a agent

a solubiliser and preferably also with alcomplexing agent. 10 In recent years estimation than focused on alkyl glucosides since these have proved to be more seasily biodegradable than other non-ionic surfactants, such as ethylene oxide adducts of fatty alcohols. US Patent Specifica-HOUSE NE-tion 3,839,318 thus describes the production of alkyl glu-72 15 M cosides and alky Foligosaccharides, such as n-octyl glucoside, n-hexyl glucoside, n-decyl glucoside, n-dodecyl glucoside, isodecyl glucoside, isoundecyl glucoside, isotriand the corresponding oligosaccharides. The United States Stationary Invention Registration H171 20 states that alkyl glucosides of formulae R(OG) and R(OG), are excellent suffactants. In these formulae, R is an the stalkylpor alkenylegroup which is branched at the second carbon atom or at a higher carbon atom, the branch being To be selected from the group methyl, ethyl, isopropyl, n-pro-25 pyl, butyl, pentyl, hexyl and mixtures thereof, provided thateR contains from about 7 to about 30 carbon atoms: on the group glucose, fructose, mannose, galactose, talose, allose, altrose, idose, arabinose, xylose, lyxose, ribose and mixtures 30 thereof; and x is 2 or more. Example 1 contains a description of the production of two product mixtures substantially made up of 2-ethylhexyl glucoside and isooctyl glu-

DE 20 36 472, EP 306 650, EP 306 651 and EP 366 652, 35 inter alia, also describe alkyl glucosides.

coside, respectively.

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Even though alkyl glucosides generally are easily biodegradable, they are only used to a limited extent in many ranges of application, such as the cleaning of hard surfaces, since they are too high-foaming and/or have too poor a cleaning power. It is therefore a desideratum to provide non-ionic surfactants which are about as easily biodegradable, but which have a better cleaning effect on hard surfaces and/or are more low-foaming than known alkyl glucosides.

According to the present invention, it has now surprisingly been found that an alkyl glucoside of formula

wherein R is an alkyl group having a total of 8-12 carbon atoms and containing 2-4 groups of formula -CH(CH₃)- in its carbon chain, G is a monosaccharide residue, and x is 1-4, is advantageously used as a surfactant in compositions for cleaning hard surfaces. The alkyl glucoside of formula I shows good cleaning and wetting properties, as well as low foaming compared with other alcohols of approximately the same chain length. In addition, the alkyl glucoside has proved to be easily degradable and have low biotoxicity. Tests have not shown any skin irritations caused by the alkyl glucosides. Preferably, there are 2 or 3 methyl groups. Compounds in which R contains 9 or 10 carbon atoms and x is 1 or 2 are especially preferred, having a good cleaning power and being comparatively easy to produce.

The compounds according to the invention can be produced in conventional manner by reacting an alcohol of formula

RCH₂OH (II)

wherein R is as indicated above, with a monosaccharide in the presence of an acid catalyst, the molar ratio of the alcohol to the monosaccharide being 2:1-80:1. The catalyst may be an inorganic or organic acid. The reaction is per. WO 94/21769 PCT/SE94/00198

formed under vacuum at 90-120°C for about 1-4 h. Conveniently, the resulting reaction mixture is first filtered and then neutralised with an organic and or an inorganic Bolive base whereupon excess alcohol is carefully removed e.g. The teach of the teach of the second section of the sectio and about saging alcohols to formula (A) Jean she produced in conwentional manner by condensing propene, butene or mixtures water from thereof, whereuponatheadle, atrigoratetramers obtained mare prolonged with a carbon atomaby the oxoprocess. The 10 resulting aldehydes may then measily be converted to the data corresponding alcohols The alcohols tobtained form a complex mixture of methyl-branched structures balthough some ethyl substituents may be present the ramount of quaternary carbon found in the carbon chain is very small, and 15. alcohols containing quaternary carbon are to be regarded as impurities not encompassed by the invention. Examples of suitable alcohols are Exxal 9, Exxal 10, Exxal 11. Exxal 12 and Exxal 13, all sold by Exxon Chemical. The monosaccharides used as reactant suitably consists of pen-20 tose and hexose. Specific examples of monosaccharides used in the production of the inventive glucosides are glucose, mannose, galactose, otalose, allose, altrose, idose, arabinose, xylose ribose and lyxose Glucose is usually pre-

- age ferred for a commercial reasons to the 25 The alkyl glucosides according to the invention are suitable for usedin compositions for cleaning hard surfaces, e.g. for degreasing such surfaces or washing up. Excellent results are obtained in the degreasing of lac
 - quered or unlacquered metal surfaces. Apart from the 30 inventive alkyl glucoside, these compositions preferably contain a water-soluble solubiliser and suitably contain a complexing agentary setting periods a quitable to a

Examples of solubilisers are alkyl ether polyalkylene glycol, such as monobutyl diethylene glycol; glycols, such 35 as diethylene glycol, dipropylene glycol and propylene glycol; alcohols, such as ethanol, propanol and isopropanol; alkyl glucosides in which the alkyl group has

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4-8 carbon atoms; and/or tertiary or quaternary amine alkoxylates in which the alkyl group, which may be straight or branched, saturated or unsaturated, has 8-20 carbon atoms and in which 6-30 mol of alkylene oxide is 5 added per mol of amine Preferably, 50-100 mol per cent of the added alkylene oxide consists of ethylene oxide, the remainder preferably consisting of propylene oxide or a mixture of propylene oxide and butylene oxide. The different alkylene coxides can be added randomly or in blocks. 10 If the cleaning composition should be exceptionally lowfoaming, the alkylene oxide chain conveniently ends with an addition of 1-5 mol of propylene oxide and/or butylene oxide. The ratiosof the solubiliser to the inventive alkyl glucoside is usually 1:10-5:1, preferably 1:3-3:1.

The complexing agent may be a conventional inorganic or organic agent, such as an inorganic phosphate or NTA, EDTA, citric acid or a polycarboxylate. The amount added may vary from nothing at all to 300% by weight of the inventive alkyl glucoside. Preferably, the quantitative 20 ratio of the complexing agent to the alkyl glucoside is

The cleaning compositions may further contain other additives, such as pH-adjusting agents, antifoaming agents, enzymes, other surfactants and scents. The com-25 positions are usually aqueous and in the form of emulsions, microemulsions or solutions.

The invention will now be further illustrated by a few Examples.

. Example 1

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An alkyl glucoside was produced by reacting 2.6 mol 30 of an alcohol (Exxal 9) of formula (II), wherein R is a $C_{\rm g}$ alkyl having a methyl substitution of about 2 (average value), with 0.4 mol of glucose in the presence of 0.015 mol of sulphuric acid as catalyst at 110°C and 70 mbar. The reaction was interrupted after 105 min. The resulting product mixture was treated by distilling off excess alcohol under vacuum. The yield was 105 g, consisting of 60%

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of C₉-branched alkyl monoglucoside, 15% of C₉-branched alkyl diglucoside and a residue of higher oligomers. The glucosides had an average degree of polymerisation (DP) of about 1.5: The structure was determined by gas chromatography, mass spectrometry and NMR.

Example 2

An alkyl glucoside was produced by reacting 7.6 mol of an alcohol (Exxal 10) of formula (Fi); wherein R is a Cg alkyl having a methyl substitution of about 2.2 (average value), with 1.2 mol of glucose of the presence of 0.015 mol of sulphuric acid as catalyst at 90-111°C and 100 mbar. The reaction was interrupted after 120 min. The reaction mixture was treated by distilling off excess alcohol under high vacuum. The yield was 278°g, consisting of 60% of monoglucoside, 12% of diglucoside and a residue of higher oligomers. The glucosides had an average DP of 1.6.

Here, 20 ml of each of the cleaning compositions

20 below, diluted with 10 parts by weight of water per part
by weight of the composition, was applied on a vertically
arranged iron sheet soiled with mineral oils, soot, salts
and clay. After application, the coated surface was rinsed
with water without any mechanical treatment.

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	Components	Со	mpos	itio	n, 8	by.	weig	ht
, • 2 5	Birth gur 1988 Alaman Burth (1997) - 1997 Stort (1997) - 1997		2	. 2. A C	В	С	D	E
	Glucoside (Example 1)	· 5-	1, 1					
. :.	¿Glucoside (Example 2)		5	2	ļ., c	٠.,٠	à i	
1	gGlucosidecA electronic	0.55	1 25	5	$d \approx$	410	(6.1)	
. 10	Glucoside, Bo to from the	4	.12 .	2452	5.5	e 100 m		. :
**	Glucoside C	• -	100	7O 1	z.n ?	ં5ે	200	
	Glucoside D	1.11	1. 40	1. 16.5	%	*257	5	l
6.1. K	Butyldiethylene glycol	er e		11 %	11 0	11	11	11
	Quaternary ethoxy-	;			LI C	- 4	ţ.	
15	lated fatty amine		37.13	2.40	y 27			
	(Berol 555)	. 4 .	4 ,:			$\gamma_1 \cdot r_2$	3 D .	
	NTA	5	5	3	3	3	- 3	5
	Water	86	86	81	81	81	81	84

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A CONTRACTOR OF THE CONTRACTOR OF THE Glucoside A = 2-ethylhexyl-O(G) $_{x}$ H

Glucoside B = isooctyl-O(G) H Glucoside C = n-dodecyl/n-tetradecyl glucoside (APG-600, Henkel)

Glucoside D = n-decyl glucoside (Lutensol GD-70, BASF) 25 wherein G = glucoside residue and x = 1.5 (average value).

The attained cleaning effect was assessed with respect to the area of the cleaned surface, as well as its actual cleanness, the figure 1 indicating no improvement and the figure 10 indicating a perfectly clean surface. The following results were obtained.

		,	
1.20]	Cleaned surface; cm ²	
5.	s koos je na na jest (one car 112 ; ser to si where, a sol t-144 ; may solt	1460 14 6 30 mp
		Charles Afrika Charles (1784) O Tiskas (1880)	กก calvad
10	B	80 48	4 6
	D	72	6
	E	0	1

The foaming of the different ready-to-use solutions was measured according to Ross-Miles ASTM D 1173-53. The following results were obtained.

20	Composition	Foam heigh	nt, mm
	Composition	Instantaneously	After 5 min
	. 1	5	0
25	2	8	0
	A	. 7	0
	В	20	3
	С	67	63
	· •	46	. 45
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It is evident from these results that the alkyl glucosides according to the invention show an excellent cleaning power and are clearly superior to alkyl glucosides having a straight carbon chain with 10-14 carbon atoms, while at the same time having an acceptable degree of foaming. The composition containing alkyl glucosides having an alkyl group with 8 carbon atoms showed an unsatisfactory cleaning power.

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1. The use of an alkyl glucoside of the general for-5 mula http://dx.

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- wherein Reis and alkylogroup having a total of 8-12 carbon atoms and containing 2-4 groups of formula -CH(CH3) in its carbon chain; G is a monosaccharide residue; and x is 1-4, as a surfactant for cleaning hard surfaces.
 - 2. Use as set forth in claim 1, characterised in that R is an alkyl group having 9 or 10 carbon 15 atoms.
 - 3. Use as set forth in claim 1 or 2, characterised in that R has a methyl substitution of 2 or 3.
 - 4. Use as set forth in claim 1, 2 or 3, charac20 terised in that G is a glucose residue.
 - 5. Use as set forth in any one of claims 1-4, characterised in that x is 1 or 2.
 - 6. Use as set forth in claims 1-5 of the alkyl glucoside of formula (I) as a surfactant in a cleaning composi-
 - 25 tion for degreasing lacquered or unlacquered metal surfaces.
 - 7. A cleaning composition, characterised in that it contains, in addition to the alkyl glucoside of formula (I), a water-soluble solubiliser and, optionally, an organic or inorganic complexing agent.
 - 8. A cleaning composition as set forth in claim 7, c h a r a c t e r i s e d in that the solubiliser consists of alkyl ether polyglycols, glycols, alcohols and/or tertiary and/or quaternary alkylamine alkoxylates.
 - 35 9. A cleaning composition as set forth in any one of claims 6-8, characterised in it contains a solubiliser in an amount of 1:3-3:1 based on the weight of

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the alkyl glucoside, and a complexing agent in an amount of 1:10-2:1 based on the weight of the alkyl glucoside.

10. An alkyl glucoside of the general formula .

 $5 \qquad \text{RCH}_2 O(G)_{x} H \qquad (.1)$

wherein R is an alkyl group having a total of 8-12 carbon atoms and containing 2-4 groups of formula -CH(CH₃)- in its carbon chain; G is a monosaccharide residue; and x is 1-4.

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INTERNATIONAL SEARCH REPORT

International application No.

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A. CLASSIFICATION OF SUBJECT MATTER		r : 6137
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B. FIELDS SEARCHED (1972)	national classification and IPC	
Minimum documentation searched (classification system followed	by classification symbols)	
IPC5: C11D, C07H		
Documentation searched other than minimum documentation to the	ne extent that such documents are included i	n the fields searched
SE,DK,FI,NO classes as above		
Electronic data base consulted during the international search (nam	ne of data base and, where practicable, searc	n terms used)
C. DOCUMENTS CONSIDERED TO BE RELEVANT		
Category* Citation of document, with indication, where ap	opropriate, of the relevant passages	Relevant to claim No.
A US, A, 3839318 (RICHARD C. MANS 1 October 1974 (01.10.74)	FIELD),	1-10
		
A DE, B2, 2036472 (ATLAS CHEMICAL 7 December 1978 (07.12.78)	INDUSTRIES INC.),	1-10
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INTERNATIONAL SEARCH REPORT

Information on patent family members

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	document arch report	Publication date	Patent family member(s)		Publication date
US-A-	3839318	01/10/74	NONE		
DE-B2-	2036472	07/12/78	GB-A- 127	5596 7516 2269	07/05/71 14/06/72 13/11/73

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